

Supporting Information

Metal-free Lewis Pair Catalysts for a One-pot Terpolymerization of Propylene Oxide, L-Lactide and CO₂

Shuxian Ye ^{1,‡}, Jiaxin Liang ^{1,‡}, Yansong Ren ¹, Shuanjin Wang ¹, Dongmei Han ⁴, Sheng Huang ¹, Zhiheng Huang ¹, Min Xiao ^{1,*} and Yuezhong Meng ^{1,2,3,4,*}

¹ The Key Laboratory of Low-carbon Chemistry & Energy Conservation of Guangdong Province/State Key Laboratory of Optoelectronic Materials and Technologies, School of Materials Science and Engineering, Sun Yat-sen University, Guangzhou 510275, China; yeshx7@mail3.sysu.edu.cn (S.Y.); liangjx27@mail2.sysu.edu.cn (J.L.); renys@mail2.sysu.edu.cn (Y.R.); wangshj@mail.sysu.edu.cn (S.W.); huangsh47@mail.sysu.edu.cn (S.H.); hzh29@mail.sysu.edu.cn (Z.H.)

² Institute of Chemistry, Henan Academy of Sciences, Zhengzhou 450000, China

³ Research Center of Green Catalysts, College of Chemistry, Zhengzhou University, Zhengzhou 450001, China

⁴ School of Chemical Engineering and Technology, Sun Yat-sen University, Guangzhou 510275, China; handongm@mail.sysu.edu.cn (D.H.)

[†] These authors contributed equally to this work.

* Corresponding authors. E-mail: stsxm@mail.sysu.edu.cn (M.X.); mengyzh@mail.sysu.edu.cn (Y.M.)

This Supporting Information contains:

Number of pages: 8

Number of Figures: 9

Number of tables: 4

Number of schemes: 2

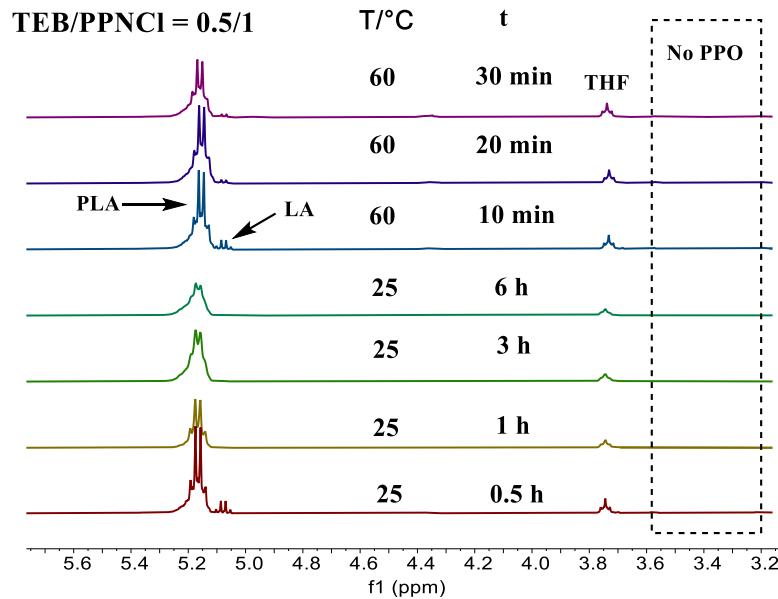


Fig. S1. ^1H NMR spectra of LA ROP at different times and temperatures (PO/LA/TEB/PPNCl = 500/100/0.5/1).

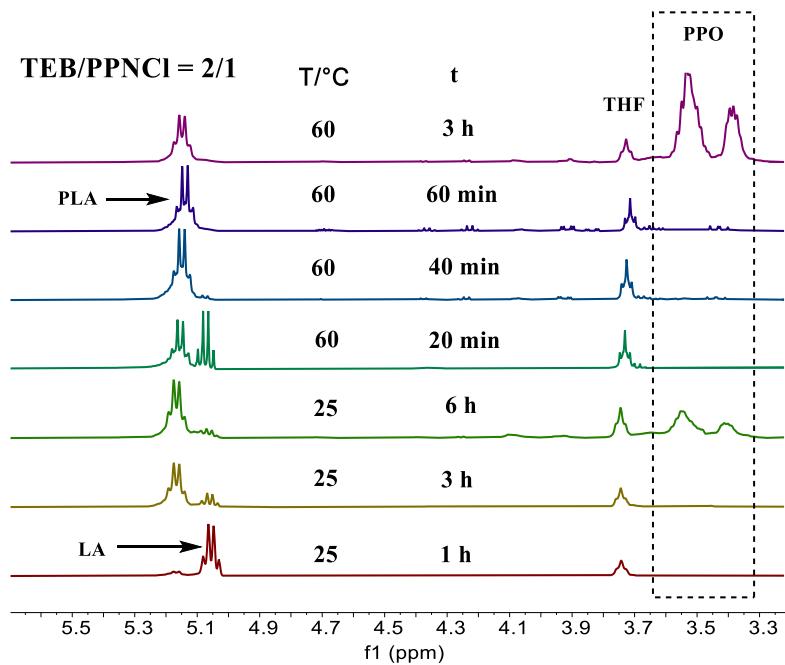


Fig. S2. ^1H NMR spectra of LA ROP at different times and temperatures (PO/LA/TEB/PPNCl = 500/100/2/1).

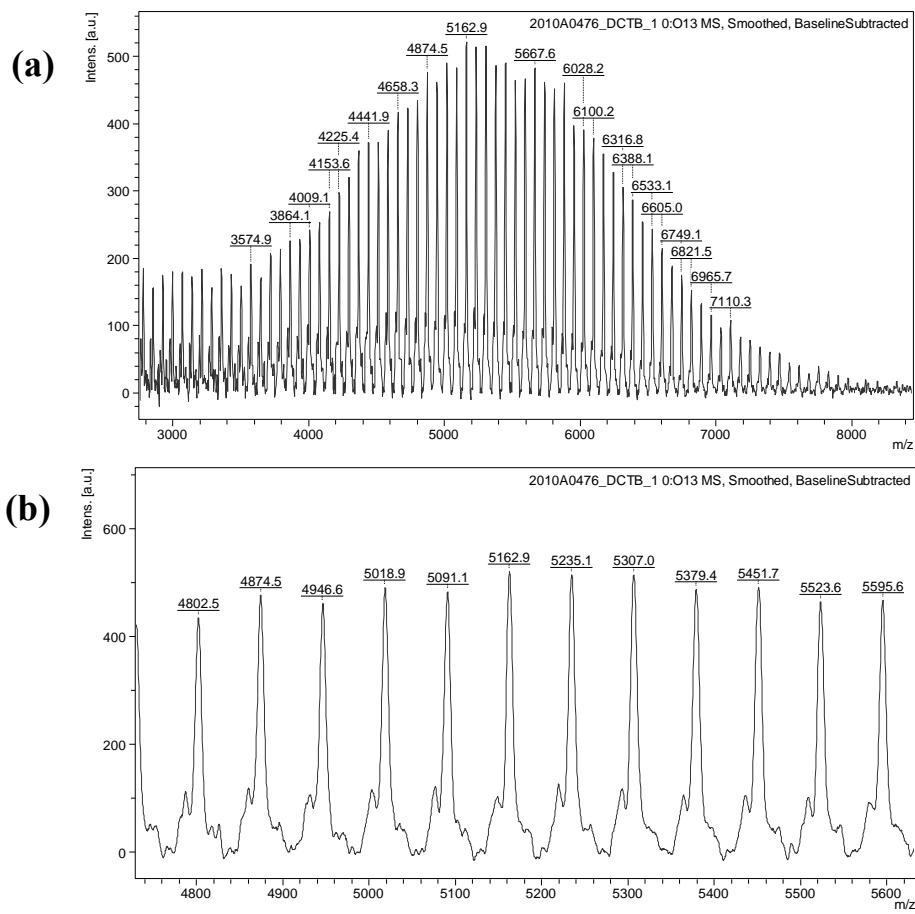
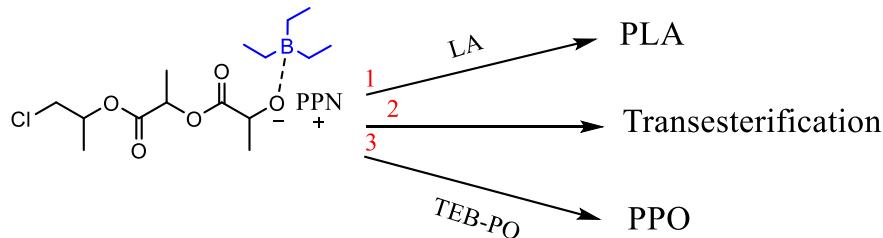


Fig. S3. MALDI spectra of PLA obtained from the ROP of LA catalyzed by TEB/PPNCl = 1/1 at 25 °C for 60 minutes: (a) initial spectrum; (b) the magnified part. The major series of the oligomers can be assigned as $A_n = (1/2LA)_nPOClNa^+$ with 72.02 interval



Scheme S1. Three kinds of reactions in the synthesis of PLA with PO as solvent.

Table S1. Synthesis of PPCLA at different temperatures ^a

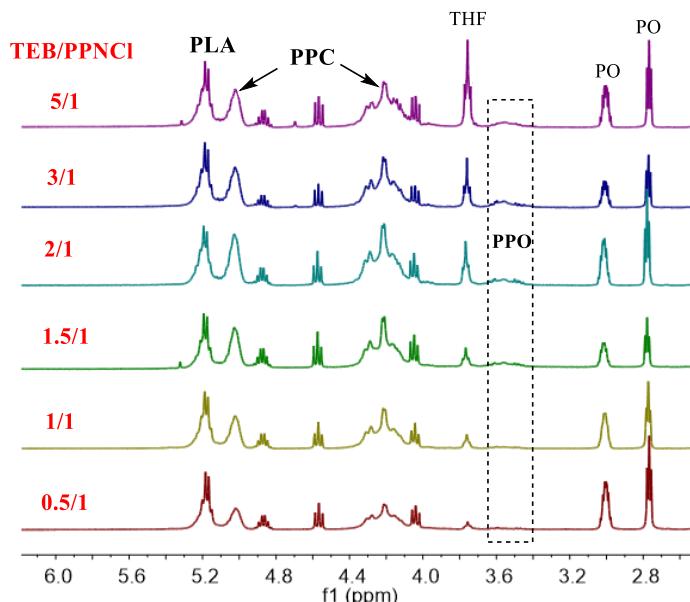
entry	temp. (°C)	select. (%) ^b	composition (mol%) ^c			M_n (kg·mol ⁻¹)/PDI ^d
			PLA	PPC	PPO	
1	60	90	19	78	3	9.5/1.60
2	70	82	16	76	8	9.2/2.26
3	80	66	17	73	9	10.9/2.30

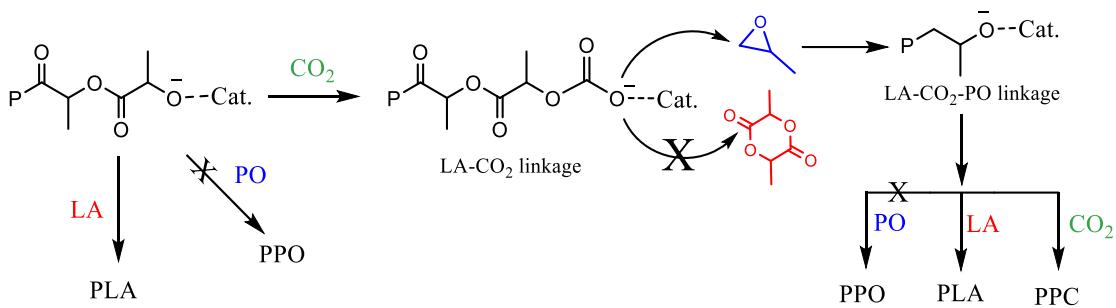
^a All polymerizations were conducted under a PO/LA/TEB/PPNCl molar ratio of 920/80/2/1 at 1 MPa CO₂ for 12 h. ^b Calculated by ¹H NMR, linear vs cyclic selectivity. ^c Calculated by ¹H NMR. ^d Determined by GPC in chloroform with polystyrene standard.

Table S2. Synthesis of PPCLA catalyzed by different TEB/PPNCl ratios ^a

entry	TEB/PPNCl	select. (%) ^b	composition (mol%) ^c			$M_n(\text{kg}\cdot\text{mol}^{-1})/\text{PDI}^d$
			PLA	PPC	PPO	
1	0.5	78	69	31	0	5.9/2.40
2	1	82	48	52	<1	8.7/2.08
3	1.5	87	38	61	1	9.3/1.84
4	2	93	33	65	2	10.1/1.58
5	3	92	32	66	2	10.6/2.16
6	5	90	31	66	3	9.1/2.35
7 ^e	0.5	80	75	25	0	6.3/2.45
8 ^f	0.5	0	-	-	-	No polymer

^a All polymerizations were conducted under a PO/LA/PPNCl molar ratio of 500/100/1 at 1 MPa CO₂, 60 °C for 4 h or under conditions otherwise mentioned. ^b Calculated by ¹H NMR, linear vs cyclic selectivity. ^c Calculated by ¹H NMR. ^d Determined by GPC in chloroform with polystyrene standard. ^e The reaction was conducted at 25 °C for 1 h first and then charged with 1 MPa CO₂ to react at 60 °C for 3 h. ^f No addition of LA; only propylene carbonate was formed.

**Fig. S4.** ¹H NMR spectra of PPCLA at different TEB/PPNCl ratios (Table S2).



No long-chain PPC can be formed because of faster backbiting rate than chain growth

Scheme S2. Proposed chain propagation process at TEB/PPNCl = 0.5/1.

Table S3. Terpolymerization of PO, LA and CO₂ catalyzed by TEB/PPNCl (2/1) ^a

entry	time (h)	LA conv. (%) ^b	select. (%) ^c	composition (mol%) ^b			$M_n(\text{kg}\cdot\text{mol}^{-1})/\text{PDI}^d$
				PLA	PPC	PPO	
1	2	61	84	78	22	0	5.8/1.47
2	4	100	89	33	65	2	10.1/1.58
3	6	100	93	30	66	4	11.4/1.80
4	8	100	93	26	69	5	12.8/1.88
5	10	100	94	24	69	7	13.6/2.23

^a All polymerizations were conducted under a PO/LA/PPNCl molar ratio of 500/100/1 at 1 MPa CO₂, 60 °C. ^b Calculated by ¹H NMR. ^c Calculated by ¹H NMR, linear vs cyclic selectivity. ^d Determined by GPC in chloroform with polystyrene standard.

Table S4. Terpolymerization of PO, LA and CO₂ catalyzed by TEB/DBU/BDM (2/1/0.5) ^a

entry	time (h)	LA conv. (%) ^b	select. (%) ^c	composition (mol%) ^b			$M_n(\text{kg}\cdot\text{mol}^{-1})/\text{PDI}^d$
				PLA	PPC	PPO	
1	1	50	84	79	21	0	2.8/1.51
2	2	83	86	59	41	0	4.3/1.51
3	4	100	88	38	60	2	8.7/1.44
4	6	100	90	30	66	4	9.2/1.64
5	8	100	88	24	67	9	12.6/2.06
6	10	100	86	24	65	10	13.2/2.32

^a All polymerizations were conducted under a PO/LA/DBU molar ratio of 500/100/1 at 1 MPa CO₂, 60 °C. ^b Calculated by ¹H NMR. ^c Calculated by ¹H NMR, linear vs cyclic selectivity. ^d Determined by GPC in chloroform with polystyrene standard.

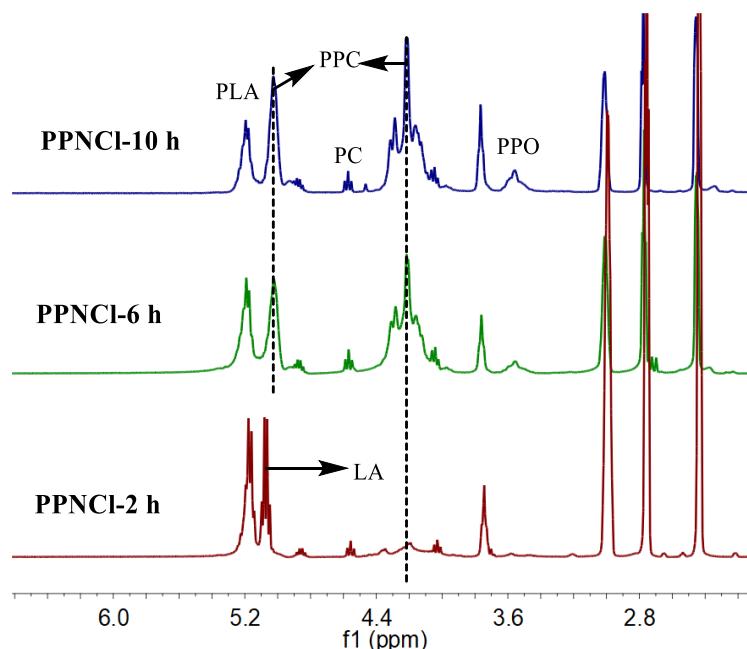


Fig. S5. ¹H NMR spectra of PPCLLA catalyzed by TEB/PPNCl at different time (Table S3).

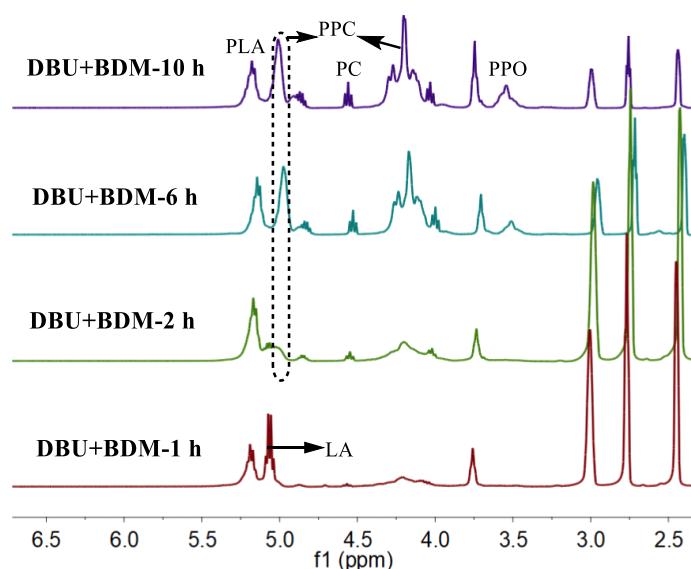


Fig. S6. ¹H NMR spectra of PPCLLA catalyzed by TEB/DBU/BDM at different time (Table S4).

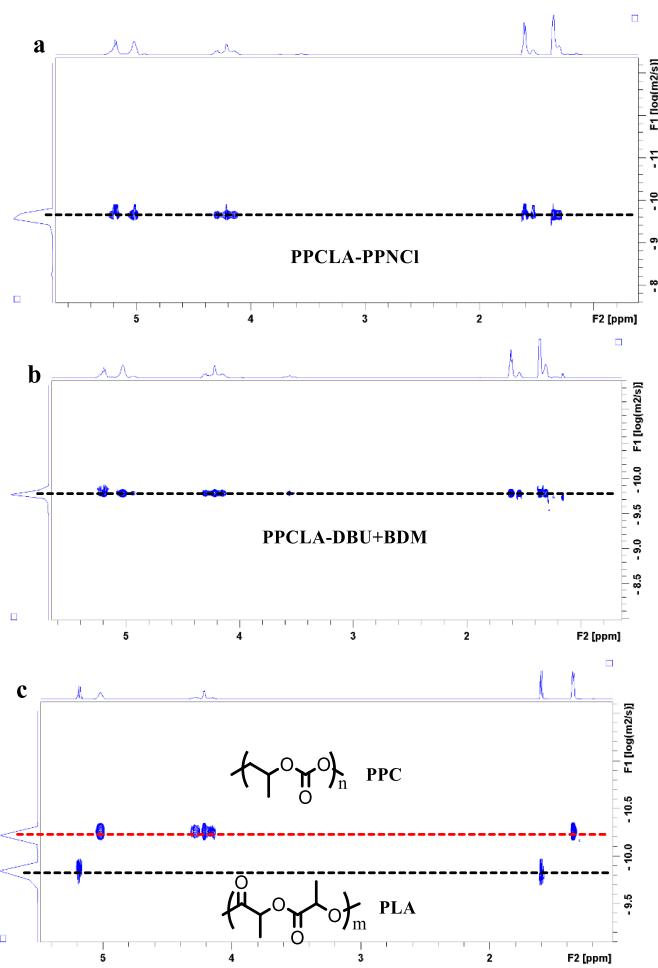


Fig. S7. DOSY spectra of (a) PPCLA-PPNCl; (b) PPCLA-DBU+BDM; (c) the mixture of PPC and PLA.

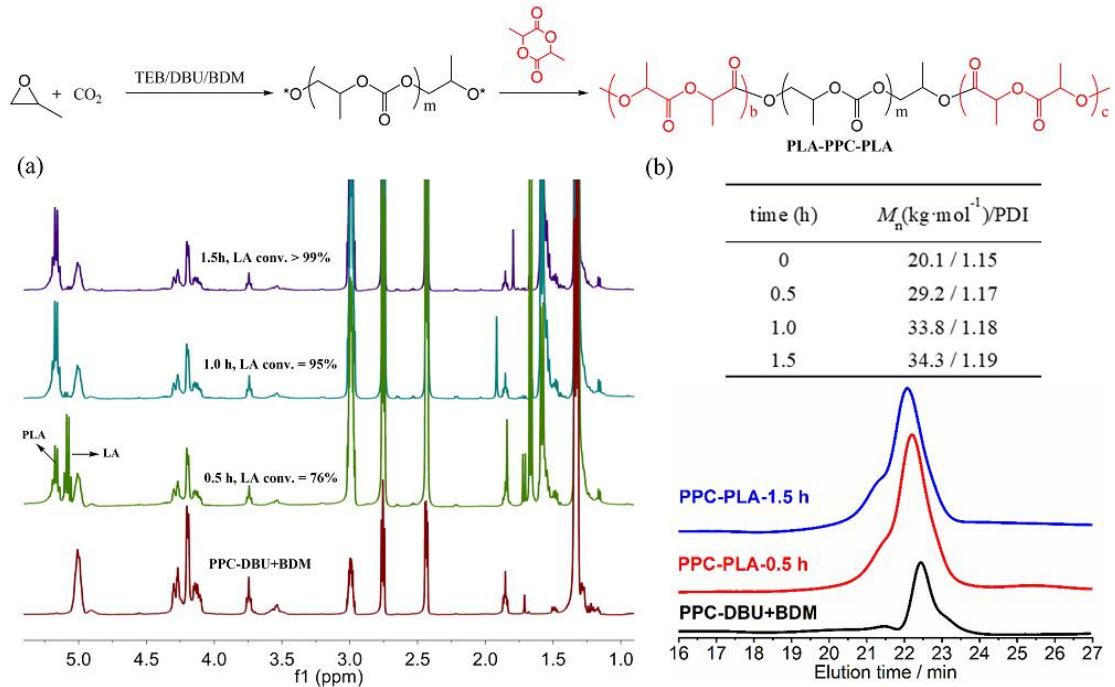


Fig. S8. Synthesis of PPC-PLA block copolymers catalyzed by TEB/DBU/BDM: (a) Evolution of ^1H NMR spectra of LA ROP with reaction time; (b) Evolution of GPC traces and MWs.

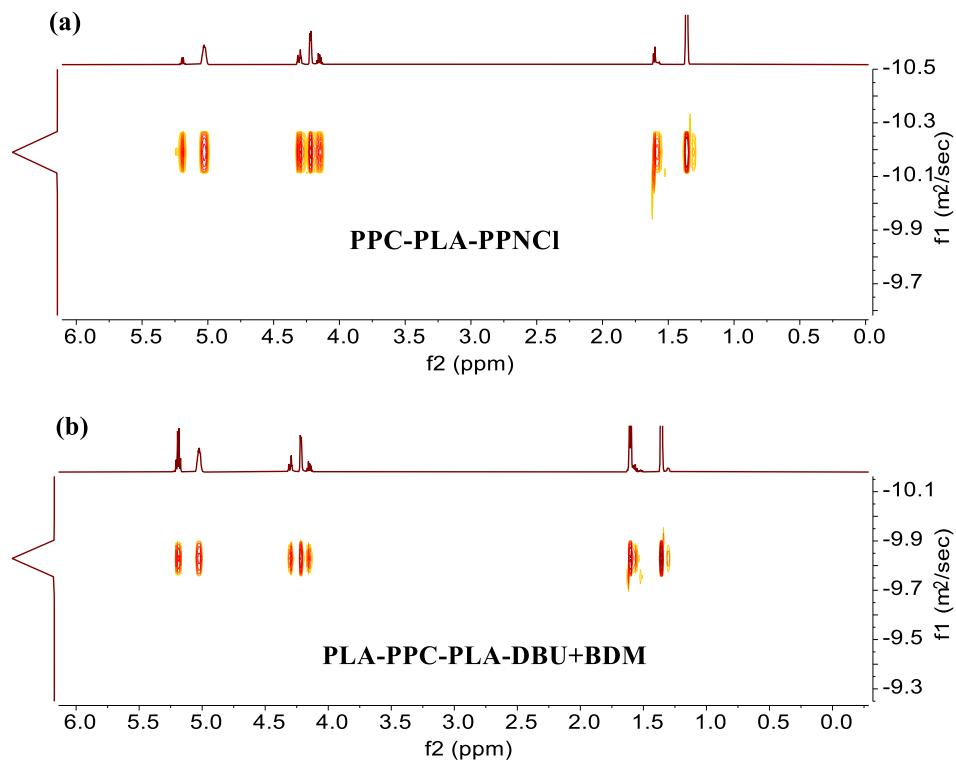


Fig. S9. DOSY spectra of (a) PPC-PLA-PPNCl; (b) PLA-PPC-PLA-DBU+BDM